Supporting Information for

"Optical crack detection and assessment using cholesteric liquid crystal elastomers" Tarik Camo, Rijeesh Kizhakidathazhath, Danièle Waldmann-Diederich and Jan P.F. Lagerwall

The molecular structures of the monomers employed in the preparation of CLCE films are illustrated in Figure 2a of the main paper. The composition of the CLCE precursor solution used in this study can be seen in Table S1. The diacrylate mesogen, 1,4-bis-]4-(3-acryloyloxypropyloxy)benzoyloxy]-2-methylbenzene (RM257) (Synthon Chemical) and the chiral dopant (3R,3aS,6aS)-hexahydrofuro[3,2-b]furan-3,6-diyl bis(4-(4-((4- (acryloyloxy)butoxy)carbonyloxy)benzoyloxy)benzoate) (LC756) (Synthon Chemical) were heated to 80°C for 5 minutes and subsequently cooled to room temperature. Following this, a tetra-functional thiol cross-linker pentaerythritol tetrakis(3-mercaptopropionate) (PETMP) (Sigma Aldrich), a dithiol monomer 2,2-(ethylenedioxy) diethanethiol (EDDET) (Sigma Aldrich), a photoinitiator 2,2-dimethoxy-2-phenylacetophenone, Irgacure 651 (Sigma Aldrich) and a dye, Sudan black B, were mixed with the solution. Finally, dipropylamine diluted to 1: 50 ratio with toluene was introduced as a catalyst for the first stage Michael addition reaction. The solution was stirred and was then applied to a substrate such as brick, polystyrene, etc., and left open for 12 hrs. Subsequently, the CLCE film was exposed to UV light at 365 nm with an intensity of 32 mW cm⁻² for 5 min at 25°C.

Table S1: Composition of CLCE system used for this study. Substance amount is defined as the given mass of a substance devided by its molecular mass.

Name	Mass (g)	Molecular mass	Substance amount	CAS Registry nr.	
		(g/mol)	(mmol)		
RM257	1.92	588.6	3.26	174063-87-7	

LC756*	0.090	966.89	0.09	223572-88-1
EDDET	0.498	182.3	2.7	14970-87-7
PETMP	0.130	488.7	0.27	7575-23-7
Irgacure 651	0.025	256.3	0.09	24650-42-8
Sudan Black	0.003	456.54	0.007	4197-25-5
Dipropylamine**	0.340	101.2	-	142-84-7
Toluene	1.060	92.1	-	108-88-3

Estimation of the cost of the CLCE precursor solution and the final CLCE coating

We approximate the density of the CLCE to 1 g/m³, and we approximate mass-% to be equal to volume-%. We assume a suitable thickness of the final CLCE coating to be 25 μ m (for the optics already some 5 μ m would be enough, but we want the CLCE layer to be reasonably durable also). Thus: the volume of 1 m² CLCE can be calculated as 10⁴ cm² area * 25*10⁻⁴ cm thickness = 25 cm³ = 25 mL ≈ 25 g. Since everything in the precursor solution except the toluene remains in the final

^{* 4.48} wt % of LC756 with respect to RM257 (to obtain a CLCE coating with a red-retroflection ground state)

^{** 0.340} g of DPA/toluene solution (DPA was diluted in toluene at a ratio of 1:100)

CLCE, we can estimate the mass of precursor solution required for covering 1 m² of CLCE as $25/0.74 \approx 34$ mL. With the cost of the precursor being about 1.67 €/mL (see Table S2), this means that the current cost of the CLCE coating is about $56 \text{ } \text{€/m}^2$. Note that this is based on the current situation, where the mesogens are speciality chemicals. If the chemical industry scales up production of these components, we can expect a significant drop in the cost.

Table S2: Cost of each component in the CLCE precursor.

Component	Mass (g)	Mass %	Price	Explanation	Cost of each
			estimate (€/g)		component in
					1 g of mixture
RM257	1.92	0.472	1.5	Ca 1.5 k€ per	0.708
				kg from China	
LC756	0.090	0.022	39	3.9k€ per 100g	0.858
				from Synthon	
EDDET	0.498	0.122	0.39	39€ per	0.048
				100mL from	
				Sigma	
PETMP	0.130	0.032	0.35	175€ for	0.011
				500mL at	
				Sigma	

Irgacure 651	0.025	0.006	1.36	68.10€ for 50g at Sigma.	0.008
Sudan Black	0.003	0.00073	4.16	104€ for 25g at Sigma	0.003
Dipropylamine	0.340	0.084	0.0252	504€ _for 20kg at Sigma	0.002
Toluene	1.060	0.260	0.106	92.50 € for 1000mL	0.028
Total	4.066	1			1.666